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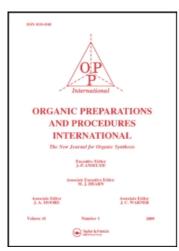
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## SYNTHESIS OF 4-ISOPROPYL-2,6-DINITROCHLOROBENZENE

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#### SYNTHESIS OF 4-ISOPROPYL-2,6-DINITROCHLOROBENZENE

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We would like to bring to your attention a convenient and efficient procedure for the preparation of 4-isopropyl-2,6-dinitrochlorobenzene by the treatment of 4-isopropyl-2,6-dinitrophenol with thionyl chloride and dimethylformamide. This method corresponds essentially to the one reported previously; 1,2 however, the yield and the quality of the product have been improved greatly by employing hexane as a solvent in the reaction. This modification provides a pure crystalline material directly from the reaction mixture without any purification and could also be utilized as a general synthesis for 4-alkyl-2,6-dinitrochlorobenzenes.

#### EXPERIMENTAL

4-Isopropyl-2,6-dinitrochlorobenzene.-Dimethylformamide (40 ml.) was added to a suspension of 4-isopropyl-2,6-dinitrophenol (452 g., 2 moles)

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in hexane (400 ml.). The reaction mixture was heated to  $40-50^{\circ}$ C, and thionyl chloride (333 g., 2.8 moles) was added dropwise over a period of 45 minutes. The reaction mixture was refluxed for 5 hours and cooled. Dilute sodium bicarbonate solution (5%) was added to neutralize the reaction mixture. The resulting mixture was stirred for thirty minutes and the product removed by filtration, washed with water, then hexane and air dried to yield 456 g. (93%) of 4-isopropyl-2,6-dinitrochlorobenzene, m.p. 61.5-64°. Anal. Calculated for  $C_9H_9ClN_2O_4$ : C, 44.58; H, 3.67. Found: C, 44.58; H, 3.81. The spectroscopic data are in accordance with the assigned structure. NMR (DMSO<sub>d6</sub>)  $\delta$  1.32 (d,6,  $\underline{J}$  = 6.9 Hz,  $\underline{CH}_3$ ), 3.21 (m,1,  $\underline{J}$  = 6.9 Hz,  $\underline{CH}$ ), 8.37 (s,2, aromatic protons); mass spectrum  $\underline{M}^+$ . 244, one chlorine atom.

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